

Lösung zum Problem Y08

1. IR-Spektrum:

3460 cm ⁻¹	vermutlich Oberton der Bande bei 1740
≈ 3000 cm ⁻¹	C-H str
1740 cm ⁻¹	C=O str

sonst wenig Information

Kein OH, NH, C≡C, C≡N, C=C

2. Massenspektrum

- m/z 258 = M⁺ · ? keine Widersprüche. M+1 nicht ablesbar
- Isotopenmuster: Br₂
- Starkes Signal bei 106: Br-Isotopenmuster
- Signale bei 179, 151: Br-Isotopenmuster
- Starkes Signal bei m/z = 55 (C₄H₇ / C₃H₃O)
- Massendifferenzen
 - Δm = 28 (258 → 230) (- CO)
 - Δm = 45 (258 → 213) (- C₂H₅O)
 - Δm = 73 (258 → 185) (?, Verlust von 28 aus 213)
 - Δm = 79 (258 → 179) (- Br)

3. ^{13}C NMR-Spektrum

Signalgruppe bei 77 ppm = Lösungsmittel.

$\delta =$	169.6	62.7	41.2	29.8	13.9	ppm
aus DEPT:	C-X	CH ₂	CH	CH ₂	CH ₃	
	O	Y	Z	(X, Y, Z: Heteroatome)		

$\Rightarrow \geq \text{C}_5\text{H}_8\text{O}$

4. ^1H NMR-Spektrum

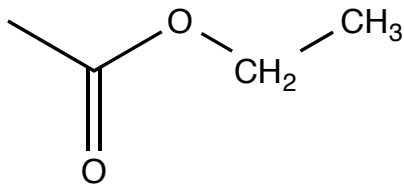
$\delta =$	4.43	4.30	3.93	3.68	1.33	ppm
Integral:	1	2	1	1	3	$\Rightarrow \text{H}_8$
Mult:	dxd	Q	Dxd	Dxd	T	
HSQC:	CH	CH ₂	CH ₂		CH ₃	
Kopplg:		┌──────────────────────────────────┐ diastereotop! └──────────────────────────────────┘				
	┌──────────┐ ┌──────────┐					
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6. Bilanz:

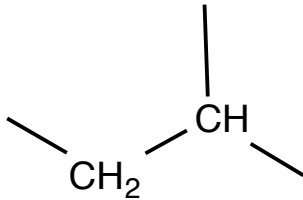
$\text{C}_5\text{H}_8\text{OBr}_2 = 242$, es fehlen 16 (1 x O; X = Y = O)

$\Rightarrow \text{C}_5\text{H}_8\text{O}_2\text{Br}_2$ Molekulargewicht = 258, DBE = 1 (C=O)

7. Strukturelemente:



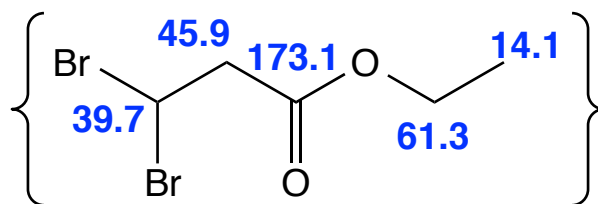
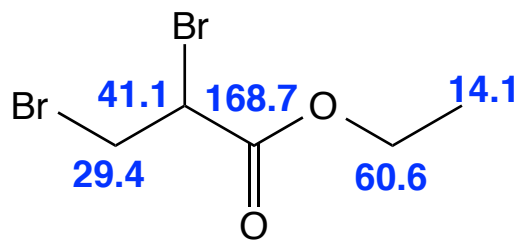
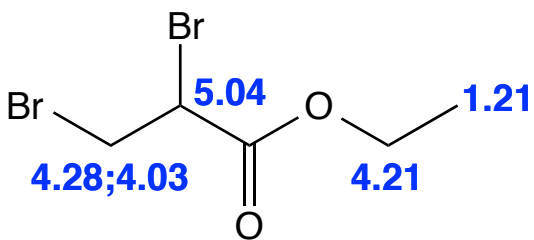
Chem. Verschiebungen
Kopplungen: CH₂-CH₃



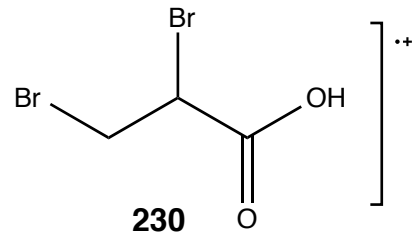
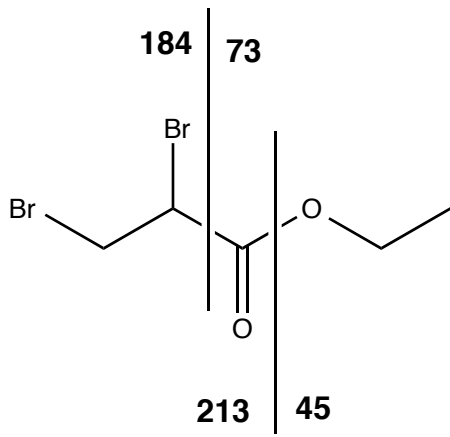
Kopplungen
CH-Gruppe: Chiralitätszentrum =>
Diastereotope CH₂

2 x -Br

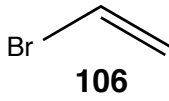
8. Verifizierung



MS-Abspaltungen:



McLafferty-Produkt



$C_3H_3O = m/z 55$